

Faculty of Metalurgy and Technology / CHEMICALL TECHNOLOGY / CHEMICAL REACTORS

Course:	CHEMICAL REACTORS								
Course ID	Course status	Semester	ECTS credits	Lessons (Lessons+Exer cises+Laboratory)					
3231	Mandatory	5	4	2+2+0					
Programs	CHEMICALL TECHNOLOGY								
Prerequisites									
Aims	Through this course student acquire basic knowledge from reactor engineering – principles and calculation techniques used to analyze and design chemical reactors, material and energy balances applied to chemical reactor design for ideal reactors								
Learning outcomes									
Lecturer / Teaching assistant	Dr Biljana Damjanovic-Vratnica, full professor MSc Dragan Radonjić								
Methodology	Lectures, tutorials, homework, midterm thesis, consultation.								
Plan and program of work									
Preparing week	Preparation and registration of the semester								
I week lectures	Introduction to Chemical Reactors and Material Balances								
I week exercises	Calculations								
II week lectures	Mole balances for ideal reactors, conversion and reactor sizing								
II week exercises	Calculations								
III week lectures	Batch reactors								
III week exercises	Calculations								
IV week lectures	Plug flow reactor								
IV week exercises	Calculations								
V week lectures	Design of ideal reactors, Process economy								
V week exercises	Calculations								
VI week lectures	Semi-batch reactors								
VI week exercises	Calculations								
VII week lectures	First midterm exam								
VII week exercises	Makeup first midterm exam								
VIII week lectures	Plug flow reactor with recycle and auto-catalytic reactions								
VIII week exercises	Calculations								
IX week lectures	Nonisothermal reactor								
IX week exercises	Calculations								
X week lectures	Nonisothermal reactor design								
X week exercises	Calculations								
XI week lectures	Nonisothermal batch reactor design								
XI week exercises	Calculations								
XII week lectures	Plug flow reactors design								
XII week exercises	Calculations								
XIII week lectures	Overall Energy Balance for Reactors								
XIII week exercises	Second midterm exam								
XIV week lectures	Nonisothermal reactor design								
XIV week exercises	Makeup second midterm exam								
XV week lectures	Reactor design for multiple reactions								



XV week ex	ercises	Calculations							
Student w	orkload								
Per week			Per semester						
<pre>4 credits x 40/30=5 hours and 20 minuts 2 sat(a) theoretical classes 0 sat(a) practical classes 2 excercises 1 hour(s) i 20 minuts of independent work, including consultations</pre>			Classes and final exam: 5 hour(s) i 20 minuts x 16 =85 hour(s) i 20 minuts Necessary preparation before the beginning of the semester (administration, registration, certification): 5 hour(s) i 20 minuts x 2 =10 hour(s) i 40 minuts Total workload for the subject: 4 x 30=120 hour(s) Additional work for exam preparation in the preparing exam period, including taking the remedial exam from 0 to 30 hours (remaining time from the first two items to the total load for the item) 24 hour(s) i 0 minuts Workload structure: 85 hour(s) i 20 minuts (cources), 10 hour(s) i 40 minuts (preparation), 24 hour(s) i 0 minuts (additional work)						
Student ob	ligations			Attending lectures, homework, midterm and final exams					
Consultations				Monday: 12-13 h.					
Literature			Recommended textbooks O. Levenspiel , Chemical Reaction Engineering, Wiley & Sons 1999. S.H. Fogler, Elements of Chemical Reaction Engineering, Prentice Hall 2005.						
Examination methods			Activity during lectures: (0 - 3 points), Activity during exercises and homework: (0 - 7 points), First midterm exam: (0 - 20 points), Second midterm exam: (0 - 20 points), Final exam : (0 - 50 points), Passing grade gets the cumulative collection at least 50 points.						
Special remarks									
Comment									
Grade:	F	E		D	С	В	А		
Number of points	less than 50 points	greater than o equal to 50 po and less than points	or bints 60	greater than or equal to 60 points and less than 70 points	greater than or equal to 70 points and less than 80 points	greater than or equal to 80 points and less than 90 points	greater than or equal to 90 points		